## **Beamont Collegiate Academy Curriculum Map**

Year: 7

Subject: Science



Intent	Implementation	Autumn 1	Autumn 2	Spring 1	Spring 2	Summer 1	Summer 2
Clarity around knowledge	Theme / topic	Working Scientifically The Particle Model (Continued into HT2)	The Particle Model Solubility and Separation	Cells and Organisation Fundamental Forces (Continued into HT4)	Fundamental Forces Core Chemistry (Continues into HT5)	Core Chemistry Reproduction	Space Science Ecology
	Key substantive knowledge	Biology:	Biology:	Biology:	Biology:	Biology:	Biology:
	Ç	N/A	N/A	Structure and Function of Living Things	N/A	Reproduction  Reproduction in	Material Cycles and Energy
				Cells and		humans (as an example of a	Photosynthesis
				Organisation		mammal), including the structure and	The dependence of almost all life on
				Cells as the fundamental unit of		function of the male and female	Earth on the ability of photosynthetic
				living organisms, including how to		reproductive systems, menstrual cycle	organisms, such as plants and algae, to
				observe, interpret and record cell		(without details of hormones), gametes,	use sunlight in photosynthesis to
				structure using a light microscope.		fertilisation, gestation and birth, to include the effect of maternal	build organic molecules that are an
				The functions of the cell wall, cell		lifestyle on the foetus through the placenta.	essential energy store and to maintain levels of oxygen and carbon
				membrane, cytoplasm, nucleus,		Health	dioxide in the atmosphere.
				vacuole,		The effects of	·
				mitochondria and chloroplasts.		recreational drugs (including substance	Interactions and interdependencies
				The similarities and		misuse) on behaviour, health and life	The interdependence
				differences between		processes.	of organisms in an

		plant and animal	Genetics and	ecosystem, including
		cells.	Evolution	food webs and insect
				pollinated crops.
		The structural	Inheritance,	
		adaptations of some	chromosomes, DNA	How organisms
		unicellular organisms.	and genes.	affect, and are
				affected by, their
		The hierarchical		environment,
		organisation of		including the
		multicellular		accumulation of toxic
		organisms: from cells		materials.
		to tissues to organs to		
		systems to organisms.		Genetics and
				Evolution
				Changes in the
				environment may
				leave individuals
				within a species, and
				some entire species,
				less well adapted to
				compete successfully
				and reproduce, which
				in turn may lead to
				extinction.
				The importance of
				maintaining
				biodiversity and the
				use of gene banks to
				preserve hereditary
				material.

Chemistry:	Chemistry:	Chemistry:	Chemistry:	Chemistry:	Chemistry:
The Particulate Nature of Matter	See HT1	N/A	Atoms, elements and compounds	See HT4	N/A
The properties of the different states of matter (solid, liquid and gas) in terms of			A simple (Dalton) atomic model.  Differences between		
the particle model, including gas pressure.			atoms, elements and compounds.		
Changes of state in terms of the particle model.			Chemical symbols and formulae for elements and compounds.		
Pure and Impure Substances			Conservation of mass changes of state and		
The concept of a pure substance.			chemical reactions.  Chemical Reactions Chemical reactions as		
Diffusion in terms of the particle model.			the rearrangement of atoms.		
Mixtures, including dissolving.			Representing chemical reactions using formulae and		
Simple techniques for separating mixtures: filtration,			using equations.  The pH scale for		
evaporation, distillation and chromatography.			measuring acidity/alkalinity; and indicators.		
			Reactions of acids with metals to		

			produce a salt plus		
			hydrogen.		
			nydrogen.		
			The Periodic Table		
			The varying physical		
			and chemical		
			properties of		
			different elements.		
			The principles		
			underpinning the		
			Mendeleev Periodic		
			Table.		
			The Periodic Table:		
			periods and groups;		
			metals and non-		
			metals.		
			How patterns in		
			reactions can be		
			predicted with		
			reference to the		
			Periodic Table.		
			The consumation of		
			The properties of metals and non-		
			metals.		
Physics:	Physics:	Physics:	Physics:	Physics:	Physics:
Matter – Physical	See HT1	Describing Motion	See HT3	N/A	Space Physics
Changes				, , ,	
		Speed and the			Gravity force, weight
Conservation of		quantitative			= mass x gravitational
material and of mass,		relationship between			field strength (g), on
and reversibility, in		average speed,			Earth g=10 N/kg,

melting, freezing,	distance and time	different on other
evaporation,	(speed = distance ÷	planets and stars;
sublimation,	time).	gravity forces
condensation.		between Earth and
	Forces	Moon, and between
Similarities and		Earth and Sun
differences, between	Forces as pushes or	(qualitative only).
solids, liquids and	pulls, arising from the	
gases	interaction between	Our Sun as a star,
	two objects.	other stars in our
Brownian motion in		galaxy, other galaxies.
gases	Using force arrows in	
Diffusion in liquids	diagrams, adding	The seasons and the
and gases driven by	forces in one	Earth's tilt, day length
differences in	dimension, balanced	at different times of
concentration.	and unbalanced	year, in different
	forces.	hemispheres.
Matter – The Particle		
Model	Forces: associated	The light year as a
	with deforming	unit of astronomical
The differences in	objects; stretching	distance.
arrangements, in	and squashing –	
motion and in	springs; with rubbing	
closeness of particles	and friction between	
explaining changes of	surfaces, with	
state and shape.	pushing things out of	
· ·	the way; resistance to	
Matter – Energy in	motion of air and	
Matter	water.	
Changes with	Forces measured in	
temperature in	newtons,	
motion and spacing	measurements of	
of particles.	stretch or	
,	compression as force	
	is changed.	

Non-contact forces: gravity forces acting at a distance on Earth and in space, forces between magnets and forces due to static electricity.  Pressure  Pressure measured by ratio of force over area – acting normal to any surface.  Balanced Forces
at a distance on Earth and in space, forces between magnets and forces due to static electricity.  Pressure  Pressure measured by ratio of force over area – acting normal to any surface.
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ratio of force over area – acting normal to any surface.
area – acting normal to any surface.
to any surface.
Balanced Forces
Balanced Forces
Opposing forces and
equilibrium: weight
held by stretched
spring or supported
on a compressed
surface.
Forces and Motion
Forces being needed
to cause objects to
stop or start moving,
or to change their
speed or direction of
motion (qualitative
only).
Change depending on
direction of force and
its size.
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Clarity around sequencing	Disciplinary knowledge  Main links across the curriculum	Scientific attitudes Experimental skills and investigations Analysis and evaluation Measurement Working Scientifically develops on the KS2 Science Curriculum and underpins all of the units that follow in KS3. The Particle Model unit underpins the Separating	Scientific attitudes Experimental skills and investigations Analysis and evaluation Measurement Separating Techniques develops on The Particle Model unit (Y7).	Scientific attitudes Experimental skills and investigations Analysis and evaluation Measurement  Cells and Organisation will be developed further in Years 7 and 9. Fundamental Forces will be developed further in the Space Science unit (Y7) and Forces and Motion (Y9).	Scientific attitudes Experimental skills and investigations Analysis and evaluation Measurement Core Chemistry will be developed further in Years 8 and 9.	Scientific attitudes Experimental skills and investigations Analysis and evaluation Measurement Reproduction and Growth develops on the Cells and Organisation unit (Y7). Reproduction will be developed further in the Plant Biology unit (Y8).	Scientific attitudes Experimental skills and investigations Analysis and evaluation Measurement Space Science develops on the Fundamental Forces unit (Y7). Ecology will be developed further in Years 8 and 9.
	Authentic cross curricular links	Techniques unit (Y7).  Maths: Graphs, mean	Maths: Graphs, mean, range	Maths: equations	Maths: equations PE: speed	Maths: equations PSHE: Sex and relationships	Geography: habitats and populations, agriculture
Vocabulary	Key words	Selected key words: Hazard, variable, qualitative, quantitative, diffusion, pressure, evaporating, condensing	Selected key words: Diffusion, pressure, evaporating, condensing, soluble, insoluble, solute, solvent, saturated	Selected key words: Organelle, mitochondria, chloroplast, force, Newton, Pascal, density	Selected key words: Atom, element, compound, pH,	Selected key words: asexual, sexual, gamete, fertilisation	Selected key words: Satellites, gravity, orbit, hemisphere, ecosystem, biodiversity, adaptation, bioaccumulation, biomagnification
Assessment	Summative assessment	End of Unit Summative Test Formative Assessment Throughout	End of Unit Summative Test Formative Assessment Throughout	End of Unit Summative Test Formative Assessment Throughout	End of Unit Summative Test Formative Assessment Throughout	End of Unit Summative Test Formative Assessment Throughout	End of Unit Summative Test Formative Assessment Throughout

Links to the	Real world:	Real world:	Real world:	Real world:	Real world:	Real world:
real world /	Covid vaccine	Forensic and	IVF, cancer, medical	Air, sea and land	Air, sea and land	Weather,
careers /	development	analytical science	imaging, disease,	travel	travel, IVF, sex and	environment
PD	Careers:	Careers:	coronavirus	Careers:	pregnancy	Careers:
	Academic researcher,	Academic researcher,	pandemic	Industrial chemist,	Careers:	Astronomer,
	industrial chemist,	industrial chemist,	Careers:	engineer,	Engineer, pilot, nurse,	atmospheric
	pharmaceutical	pharmaceutical	Chemist,	pilot	gynaecologist,	scientist, ecologist,
	scientist	scientist	microbiologist,		fertility specialist,	environmental
			zoologist, botanist,		other NHS careers	biologist, zoologist
			NHS careers			